

Elemental Analysis in Soil and Fertilizer

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Introduction



O₂, CO₂ H₂O Carbon (C) Oxygen (O) Hydrogen (H)

Major nutrients

Nitrogen (N) Phosphorus (P)

Potassium (K)

Ca

Fe)

N

Mg

P

Secondary nutrients

Calcium (Ca) Magnesium (Mg) Sulfur (S) MicronutrientsBoron (B)Copper (Cu)Cobalt (Co)Iron (Fe)Chromium (Cr)Manganese (Mn)Selenium (Se)Molybdenum (Mo)Vanadium (V)Zinc (Zn)Sodium (Na)Nickel (Ni)Silica (Si)

Toxic

Arsenic (As) Cadmium (Cd) Mercury (Hg) Lead (Pb) Copper (Cu) Chromium (Cr)



Introduction





Mg



Nitrogen (N) Phosphorus (P) Potassium (K)







Soil, plant, and fertilizers

Animal feed

Human food

Soil

- The differentiation of Total Carbon (TC) and Total
 Organic Carbon (TOC) → evaluates the quality of soils
- Environmental protection → agricultural land, construction sites, playgrounds, forests, and gardens, as well as wastelands

Fertilizer

In the production process, the elemental composition of fertilizers is periodically monitored for their characterization.

- Raw materials
- Finish products



Secondary nutrients

Calcium (Ca) Magnesium (Mg) Sulfur (S) Boron (B)Copper (Cu)Cobalt (Co)Iron (Fe)Chromium (Cr)Manganese (Mn)Selenium (Se)Molybdenum (Mo)Vanadium (V)Zinc (Zn)Sodium (Na)Nickel (Ni)Silica (Si)

Micronutrients

Introduction









Toxic elements	Major uses and sources of soil contamination
Arsenic (As)	Pesticides, plant desiccants, animal feed additives, coal and petroleum, mine tailings and detergents
Cadmium (Cd)	Electroplating, pigments for plastics and paints, plastic stabilizers and batteries, fertilizers
Chromium (Cr)	Stainless steel, Chrome-plated metal, pigments and refractory brick manufacture
Lead (Pb)	Combustion of oil, gasoline and coal: Iron and steel production
Mercury (Hg)	Pesticides, catalysts for synthetic polymers, metallurgy, thermometers
Nickel (Ni)	Combustion of coal, gasoline, and oil; alloy manufacture, electroplating, batteries

Soil: Regulation in Thailand

Element	Concentration (mg/kg)	Element	Concentration (mg/kg)
As	6	Hg	22
Cd	67	Ni	436.5
Cr	17.5	Mn	1710
Cu	2920	Se	365
Pb	400		

ประกาศคณะกรรมการสิ่งแวดล้อมแห่งชาติ เรื่องกำหนดมาตรฐานคุณภาพดิน

Fertilizers: Regulation in Thailand

Element	Concentration (mg/kg)
As	50
Cd	5
Cr	300
Cu	500
Pb	500
Hg	2

ประกาศกรมวิชาการเกษตร เรื่อง กำหนดเกณฑ์ปุ๋ยอินทรีย์

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Techniques routinely used for soil analysis

	Organic element analyzer	Discrete analyzer	IC	Accelerated Solvent Extraction	GC & GC-MS	X-ray Fluorescence	ICP-OES & ICP-MS	Plasma core at 10,000 K
Nutrient Analysis	•			_	_		•	heated zone
Metal Contaminants	-			—		•		quartz
Inorganic Anions	-	—		_	—	-	—	Auxiliary Ar flow Nebulized sample
Organic Contaminants	—	—	_	•	•	—	_	ppb to %



- The Organic Elemental Analyzer is used for Carbon, Hydrogen, Nitrogen, Sulphur and Oxygen analysis
- The ICP-OES and ICP-MS are used to provide information of major plant nutrients (N, P, K), secondary plant nutrients (Ca, S, Mg), micronutrients such as B, Mn, Fe, Cu, Zn, Mo and Se, also the toxic elements (As, Cd, Pb and Hg)

What is Organic Elemental Analyzer?

Elemental characterization for agricultural purposes gives information useful for determining agronomy management plans. The determination of Nitrogen and carbon, Total Organic Carbon (TOC), and sulfur enables to characterization of

- Soils
- Leaves, Plants, Crops, and other materials
- Fertilizer



Based on the combustion of the sample. Upon combustion, the sample generates uniform compound gases of the elements C, H, N, and S. These combustion products are measured using gas chromatography and thus the ratio of the elements in the original sample is determined. C, H, N, and S can all be determined simultaneously whereas O by pyrolysis.



Quantification of the sample Quantitative oxidation of the sample Reduction of combustion gases Separation of the oxidation gases Generation of signal

Weighting Combustion Reduction Chromatography Detection



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• Introduced into the combustion reactor by and auto sampler









- Inserted in the special furnace heated at 950 °C
- A small volume of pure oxygen is added to the system and helps to burn the sample





 $O_2 + Sn \longrightarrow SnO_2 + 1800 \,^{\circ}C$









CuO Convert any CO to CO₂, N to, NOx, S to SOx, and H_2 to H_2O







- Reduction "using Copper" converting the sample into element gases
- Reduces NOx to N₂ and SOx to SO₂ and removes the excess O₂







• A separation column and TCD detector allows the user to determine elements



















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Sample Preparation

Most materials are often non-homogeneous, and an appropriate homogenization procedure is required to obtain high-quality data. The reproducibility of results is independent of the sample size but strongly dependent on the sample homogeneity.



TOC was determined after removing carbonate minerals by acidification of the sample with HCI



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Agronomy: Soil NCS and TOC





Agronomy & Animal Feed: NCS





Fertilizer

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Nitrogen determination of fertilizers according to AOAC 993.13 Official Method : applicable to the determination of 1 - 67 % total nitrogen content in liquid and solid fertilizer materials



Autosampler for liquids Autosample for solids N2 20 mm Quartz Woo Oxygen 50 mn Quartz Soda Lime 20 m luartz Woo 20 m Quartz Woo 180 mm Molecular Sieves 20 mr Quartz Nool 350 mm Copper 40 mm Reduced 20 mm Silica Gel Quartz Woo Oxidation 130 mm Catalyst 50 m GC Column Oxidation Reduction TCD Detector EagerSmart Data Handling Software

*ประกาศกระทรวงเกษตรและสหกรณ์ เรื่อง กำหนดกรรมวิธีการตรวจวิเคราะห์ปุ๋ยเคมี







Agronomy and Marine Science

Application		Official Method
a	Official Italian Method on Soils Analytical Chemistry (Gazzetta Ufficiale)	Method 248, 1999. Nitrogen, Carbon and Organic Carbon in Solis
1	ADAC (Association of Official Analytical Chemists)	Official Method 993.13. Ntrogen (Total) in Fertilizers 2.4.02
	ISO 10694, 1995 UNE 77321:2003	Soil Quality – Determination of Organic and Total Carbon After Dry Combustion (elementary analysis)
	ISO 13878, 1998 UNE 77325:2003	Soil Quality – Determination of Total Nitrogen Content by Dry Combustion (elemental analysis)
	UNE 77325:2003	Soil Quality – Determination of Total Sulfur by Dry Combustion
	UNI EN 13654-2	Soil Improvers and Growing Media. Determination of Nitrogen by Combustion Method
	Official Italian Method on Soils Analytical Chemistry (Gazzetta Ufficiale)	Method 146, 1998 Nuove Norme per la Discipina Dei Fertilizzanti (New regulations for fertilizer's control)
Carlos and	EPA (Environmental Protection Agency)	Method 440.0, 1997 Determination of Carbon and Nkrogan in Sediments and Particulates of Estuarine/Coastal Waters using Elemental Analysis

Applications: Access more applications: become the reference laboratory in your field



Petrochemistry: gasoline, diesel, lubricants, oils





Environmental and Marine Science: Wastewater, Particulate matter, Biomass, Plankton



Food: cheese, chocolate, cocoa, flours, pasta, starch, juice, wine, milk, yoghurt









Materials: paper, cotton, textile fibers, polyethylene film



Organic Chemistry & Pharm





Techniques routinely used for soil analysis



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The ICP-OES and ICP-MS are used to provide information of major plant nutrients (P, K), secondary plant nutrients (Ca, S, Mg), micronutrients such as B, Mn, Fe, Cu, Zn, Mo and Se, also the toxic elements (As, Cd, Pb and Hg)

What is ICP-OES?



What is ICP-OES?





3% TDS

What is ICP-MS?



ICP-MS is an elemental analysis technique and uses an argon (Ar) plasma - the ICP - to convert the sample into ions that are then measured using a mass spectrometer





Select the right techniques for metal analysis in your lab



Select the right techniques for metal analysis in your lab





Considers for instrument selection



- ✓ ICP-MS offers high dynamic range and lowest limits of detection
- ✓ ICP-OES has highest matrix tolerance
- ✓ GFAAs offers sensitivity, best for few elements
- ✓ Flame AA offers fast analysis, and economical sol.





 AAs
 ICP-OES
 ICP-MS

 Low (<5)</td>
 Image: Matrix
 High (>15)

 One matrix
 Sample Matrix
 Difference matrix

 ppm
 Detection limits
 ppb or lower

Sample throughput

High (>100)

Low (<20)

Elemental analysis workflow



Data







Sample preparation





Sample dilution

Measure weights and

volumes with accuracy.



Sample introduction

Sample analysis



Be aware of contamination sources.



Use ultrapure water.





Minimize handling and transfer steps.



Use high-purity reagents.



Apply proper skill, technique, and attention to detail.



Elemental analysis workflow



Data



Standard preparation



Sample preparation



Sample dilution



Sample introduction

Sample analysis

Acid Leaching





Block Acid digestion

Wet digestion : Hot plate



Microwave digestion



Dry ashing





Standard Methods



Method 200.7: Determination of Metals and Trace Elements in Water and Wastes by Inductively Coupled Plasma-Atomic Emission Spectrometry



EPA Method 6010D (SW-846) Inductively Coupled Plasma - Atomic Emission Spectrometry

Hazardous Waste Test Methods / SW-846

EPA Method 200.8 Determination of Trace Elements in Waters and Wastes by Inductively Coupled Plasma-Mass Spectrometry

EPA Method 6020B (SW-846) Inductively Coupled Plasma - Mass Spectrometry, part of Test Methods for Evaluating Solid Waste, Physical/Chemical Methods

Sample preparation techniques



Hot plate acid digestion



- Simple and inexpensive set-up involving the use of commonplace \checkmark laboratory apparatus and a hot plate
- Procedures are standardized and uncomplicated \checkmark
- Higher sample sizes (e.g., > 1 gram) are possible, which may be \checkmark required for multiphasic, heterogeneous samples



Numerous sample handling and transfer steps.



Overall inefficiency.

Hot block acid digestion



- Reduced sample handling and transfers
- Exposure to contamination is reduced.
- Elimination of issues associated with glassware.

(adsorption of analytes to container walls or leaching of elemental impurities to the sample solution, are eliminated)

X However, the duration of the digestion process is still extensive, reagent consumption is high, and exposure to contamination from the atmosphere may be possible since it is an open system.

Sample preparation techniques - Microwave digestion

- Microwaves are a form of electromagnetic radiation with relatively low energy. It promotes the rotation of specific molecules in a reaction mixture, this rotation results in increased molecular collision and the generation of heat
- Combining the sample matrix and acids (HCI, HNO₃, H₂SO₄, and H₂O₂) in a pressurized close container and elevating the solution past the boiling point of the acid
- Temperatures in the range of 200-260 °C
- Typical microwave digestion takes just 20-40 minutes

Microwave heating



✓ Speed of digestion

- ✓ Quality digestion
- Reduced exposure to contamination
- ✓ Reduced reagent consumption
- ✓ Retention of analyte

Key considerations	Open vessel a	cid digestion	Closed vessel acid digestion
	Hot plate	Hot block	Microwave
Initial investment	\$	\$\$	\$\$\$
Ease of set-up	Easiest	Easier	Easy
Consumables	N/A	Required	Optional
Maintenance and cleaning	High	Low	Medium
Sample handling	Highest handling	Lowest handling	Medium handling
Contamination exposure	Highest risk	Medium risk	Lowest risk
Reagent consumption	High	High	Low
Retention of analyte	Lowest	Medium	Highest
Digestion quality	Low	Medium	High
Batch size	Lowest	Highest	Medium
Digestion time	Hours	Hours	Minutes
Sample throughput	Lowest	Medium	High
Recommended for ultra-trace elemental analysis	Not recommended	Recommended	Highly recommended
Overall efficiency	Low	Medium	High



Soil, Sediment and Solid waste

- Nutritional elements: Al, Ba, Ca, Fe, K, Mg, Mn, P and Zn
- Trace elements: Cd, Co, Cr, Pb, Mo, Ni, V, etc.
- EPA Method 6010D (SW-846) ICP-OES





ThermoFisher

Fast, accurate, and robust analysis of

Fertilizer (AOAC 2017.02)

Elemental Categories in Fertilizer

+ HCI

Class	Elements
Primary Nutrients	N, P, K
Secondary Nutrients	Ca, Mg, Fe, Mn, Na, Cu, Zn, Mo, B, S,
Micronutrients	Al, Co, V, Se, Ni
Harmful trace elements	As, Cd, Cr, Pb

Regulatory Limits of Elements in Fertilizers in Thailand

Element	Regulatory Concentration (mg/kg)
As	50
Cd	5
Cr	300
Pb	500

thermoscientific The analysis of nitrogen, other nutrient and toxic elements in fertilizers using the Thermo Scientific ICAP 7400 IOP-OES

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100 mL with DI water

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Cannabis / Kratom / Herbal



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by Thermo Fisher Scientific

Cannabis / Kratom / Herbal







Acid digestion & Analysis by ICP-MS



Weight 0.25 - 0.5 g

Raw mat.	HNO3	HCI	H2O2
Flower	\checkmark	\checkmark	\checkmark
Leave	\checkmark	\checkmark	
Seed	\checkmark		\checkmark
Root	\checkmark		\checkmark
Products	\checkmark	\checkmark	\checkmark

Add conc. HNO₃, HCl₁H₂O₂



Microwave digestor









Qtegra Intelligent Scientific Data Solution (ISDS) Software



Thermo Scientific Qtegra ICP-MS software for 21 CFR Part 11 compliant labs

Step	Temperature (°C)	Heat time (min)	Hold time (min)
1	160	5	5
2	200	5	25

Applications: Access more applications: become the reference laboratory in your field



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